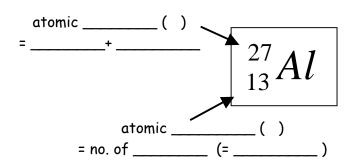
### Note Taking Guide - Program 1501

#### Nucleons:

- are particles occupying the \_\_\_\_\_
- consist of \_\_ charged \_\_\_\_\_ and \_\_\_\_
- have almost 2000 times the \_\_\_\_\_ of \_\_\_\_
- are made up of \_\_\_\_\_ and \_\_\_\_

Nuclear Notation:



 $^{27}_{13}Al$  has \_\_\_\_\_ protons and \_\_\_\_\_ neutrons, for a total of \_\_\_\_\_ nucleons.

 $^{28}_{13}Al$  has \_\_\_\_\_ protons and \_\_\_\_\_ neutrons, for a total of \_\_\_\_ nucleons.

<u>isotopes:</u> atoms of \_\_\_\_\_ element with different numbers of \_\_\_\_ (different \_\_\_\_ )

## Forces Acting on Nucleons:

# strong forces -

- forces of \_\_\_\_\_\_ between nucleons
- are independent of the \_\_\_\_\_ of the nucleon
- are short range (exist only between \_\_\_\_\_ neighbors)

## electrical force -

- force of \_\_\_\_\_ between \_\_\_\_ charged protons
- are long range

#### When are nuclei unstable?

- a. large nuclei (Z > 82) electrical forces of \_\_\_\_\_ are greater than strong forces of \_\_\_\_\_
- b. wrong neutron: proton \_\_\_\_\_

# A radioactive isotope:

- has an \_\_\_\_\_ nucleus
- spontaneously emits a \_\_\_\_\_ and \_\_\_\_ into another \_\_\_\_

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| transmutation – changing into another . | through radioactive |
|---|---------------------|
|   |                     |

# Types of Radioactive Emission:

|       | Symbol | Composition | Stopped By - |
|-------|--------|-------------|--------------|
| alpha |        |             |              |
| beta  |        |             |              |
| gamma |        |             |              |

# Nuclear Equations:

alpha decay:

$$^{214}_{84}Po \rightarrow He +$$

beta decay:

$$^{209}_{82}Pb \rightarrow e +$$

During beta decay, a neutron changes into a proton and an \_\_\_\_\_

$$_0^1 n \rightarrow _1^1 p +$$

half-life:

- the \_\_\_\_\_ it takes for \_\_\_\_ the \_\_\_\_ of a radioactive sample to \_\_\_\_
- ranges from a fraction of a \_\_\_\_\_ to billions of \_\_\_\_
- is \_\_\_\_\_ affected by \_\_\_\_\_ conditions

On back, list some dangers and uses for radioactivity, and Show What You Know.