1.	A property that depends on the number of solute particles (concentration) is said to be a property.
2.	List 5 examples of colligative properties: a)
	b)
	c)
	d)
	e)
3.	The pressure of a vapor in with its liquid is referred to as vapor pressure. It is a measure of the tendency of the molecules of a liquid to escape into the phase.
4.	The boiling point of a liquid is the temperature at which its is equal to
5.	When a solid is dissolved in a liquid, the vapor pressure of the liquid is because the surface area from which molecules can evaporate is reduced.
6.	What colligative property is responsible for antifreeze keeping water from freezing in a car's cooling system?
7.	What colligative property is responsible for antifreeze keeping water from boiling over during the summer in a car's cooling system?
8.	What colligative property causes ice to melt on the street after salt has been spread on it?