

## Main Ideas, Key Points, Questions:

After watching the video segment, write down key points, main ideas and big questions.

## NOTE-TAKING GUIDE UNIT 4, SEGMENT F

Name

Date:

## **Objective(s):**

- To analyze data from the sugar/salt melting investigation.
- To explore the career of a chemical engineer.

## Notes:

During the video segment, use words, phrases or drawings to take notes.

Summary:

After watching the video segment, write at least three sentences explaining what you learned. You can ask yourself: "If I was going to explain this to someone else, what would I say?"



QUESTIONS TO CONSIDER: UNIT 4, SEGMENT F Name:

Date:

After watching the video and performing any associated labs and/or experiments, you should be able to answer the following:

- 1. In the sugar/salt melting lab, which material melted at the lowest temperature? Which material has the strongest intermolecular forces?
- 2. If a material is made of molecules, does it probably have covalent or ionic bonds? Explain.
- 3. When sugar melts and becomes liquid, is it still sugar? Explain.
- 4. Does melting sugar cause the breaking of intramolecular bonds? Explain.
- 5. Why is it so difficult to melt an ionic compound like salt?
- 6. What is the job of a chemical engineer?
- 7. How can molecules be manufactured so they can conduct electricity?
- 8. How can molecules be manufactured so they are flexible?