

# NOTE-TAKING GUIDE UNIT 2, SEGMENT E

Name:

Date:

## Main Ideas, Key Points, Questions:

After watching the video segment, write down key points, main ideas and big questions.

#### Objective(s):

- To compare and contrast physical changes and chemical changes in matter and identify examples.
- To differentiate chemical changes from phase changes in matter.
- To explain the properties of mixtures.

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#### **Summary:**

After watching the video segment, write at least three sentences explaining what you learned. You can ask yourself: "If I was going to explain this to someone else, what would I say?"



### QUESTIONS TO CONSIDER: UNIT 2, SEGMENT E

Name:

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After watching the video and performing any associated labs and/or experiments, you should be able to answer the following:

1.	Chemical changes occur when a new chemical substance has been formed. How is a chemical change different from a physical change?
2.	Explain how you know that flammability is a good example of a chemical change.
3.	How is oxidation of copper, seen on the Statue of Liberty, similar to flammability and how is it different from flammability?
4.	How do we know that cooking pancake batter to make pancakes is definitely a chemical change?
5.	The glass hand bubbler holds a red liquid. When it heats up, do you see a chemical change or a physical change? How do you know?
6.	Draw a bowl full of different colored candies. Explain why this bowl should be called a mixture