

1. Density describes the relationship between the \_\_\_\_\_ and \_\_\_\_\_ of a sample of a substance.
2. The most common units for density are \_\_\_\_\_ and \_\_\_\_\_.
3. The density of water is \_\_\_\_\_.
4. A small beaker contains 55.0 mL of water. What would be the mass of the water?
5. Object A has a mass of 500 g and a density of  $5.0 \text{ g/cm}^3$ . Object B has a mass of 650 g and a density of  $6.5 \text{ g/cm}^3$ .
  - a. Which object would displace the most liquid? EXPLAIN.
  - b. Could the two objects be made of the same substance? EXPLAIN.
6. The density of silver is  $10.5 \text{ g/cm}^3$ . What will be the volume of a piece of silver having a mass of 31.5 g?
7. A sample of iron occupies a volume of  $10.0 \text{ cm}^3$ . If the density of iron is  $7.9 \text{ g/cm}^3$ , what is the mass of the sample?
8. Iron is sold in sheets that are 2.0 cm thick and 120.0 cm wide. You want to buy a sheet of iron with a mass of 50,000.0 g. What must be the length of the sheet of iron?